

Your Roll No.....

BSc.-MSc. Integrated Program (Chemistry)
Semester -II Examinations-2018
Paper Code- BCH-CC4
Paper Title: Status of matter and Ionic Equilibrium

Time: Three Hours

Max Marks: 75

(Write your Roll No. On the top immediately on receipt of question papers)

Attempt all questions. All parts of a question must be attempted in continuation. Start answer of a fresh question on a new page. Cite examples and draw labeled diagrams wherever necessary.

Q1. Answer very briefly of the following

[10 x1.5= 15]

- a. What is root mean square velocity?
- b. What is surface tension?
- c. Define unit cell of lattice?
- d. Define pH.
- e. Write down the units of Vander waal's constant.
- f. What is viscosity?
- g. What is the nature of solid?
- h. What is degree of ionization?
- i. What is Boyle's temperature?
- j. What is common ion effect?

Q2. Attempt ANY THREE of the following:

[3x5=15]

- A. What are the various postulates of the kinetic theory of gas?
- B. Derive kinetic gas equation.
- C. Explain compressibility factor.
- D. Explain the behavior of real gases on the basis of Vander waal's equation.
- E. How will you determine critical constant experimentally?

Q3. Attempt ANY THREE of the following:

[3x5=15]

- A. Explain the structure of liquids.
- B. Explain the XRD of Liquids.
- C. How will you determine surface tension of a liquid?
- D. What are the factors that affect viscosity of liquids?
- E. Explain the cleansing action of soap.

Q4. Attempt ANY THREE of the following:

[3x5=15]

- A. What is the law of constancy of interfacial angle?
- B. Explain the miller indices.

P.T.O.

- C. Explain Bragg's law.
- D. What are the types of defect found in crystal lattices?
- E. Explain the diffraction patterns of CsCl & NaCl.

Q5. Attempt **ANY THREE** of the following:

[3x5=15]

- A. What are the factors affecting degree of ionization?
- B. Why group iv cations do not precipitate on passing H_2S gas in presence of HCl ?
- C. What is the buffer index and buffer capacity?
- D. Derive Henderson equation.
- E. What are the various applications of Henderson Equation?

-----0-----